Elevated Temperature Deformation Behavior of Dispersion-Strengthened Al and Al-Li-Mg Alloys

Jane Minay, Richard Dashwood, and Henry McShane

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A model describing the behavior of dispersion-strengthened aluminum alloys, when subjected to elevated temperature plastic deformation, is presented. The aims are twofold:

• **to use the model for extrapolation of laboratory data to predict behavior under service conditions where the strain rate is extremely low** $(<10^{-9} \text{ s}^{-1})$; and

• **to design and fabricate materials having specific elevated temperature properties based on microstructural predictions from the model.**

The results of constant strain-rate compression tests covering a range of temperatures from 250 to 550 ^oC and strain rates of 5×10^{-5} to 10^{-1} s²¹ are presented in conjunction with microstructural investigations **using transmission electron microscopy (TEM) and x-ray diffraction. Materials mechanically alloyed with (a) no dispersoids, (b) 23 nm radius TiO2 dispersoids, and (c) 10 nm diameter Al2O3 dispersoids have been** studied. The effect of varying the volume fraction of the $TiO₂$ dispersoids and adding alloying additions **of Mg and Li to the matrix Al have been investigated. In addition, the TiO₂ particles are shown to have** reacted to form Al₃Ti. An adaptation to the detachment model of Rösler and Arzt has been proposed to **account for the behavior of these types of materials and to enable accurate prediction of deformation behavior at elevated temperatures and low strain rates.**

Lurgy routes enable the formation of finely dispersed phases
that are resistant to coarsening that cannot be produced by
conventional ingot metallurgy. Much of the work has centered
on rapidly solidified powders where disp at 315 °C^[1]), their use is limited by the hot workability. The processing step after solidification due to microstructural coars- of mechanical properties from physically based input parame-

The process of mechanical alloying overcomes the problems morphology, temperature, stress, and strain rate.

of high-temperature stability as dispersoids that are thermally Following transmission electron microscopy

Keywords aluminum alloys, dispersion strengthened, high The results presented are derived from mechanically alloyed temperature, mechanical behavior aluminum alloys strengthened with ultrafine (10 to 25 nm) thermally stable ceramic dispersoids. These materials have been **found to have excellent room-temperature and elevated temper- 1. Introduction** and $\frac{1}{2}$ at 20 °C an The demand for lightweight aluminum-based materials for
medium strengthened for lightweight aluminum-based materials for
medium strengthened temperature aerospace applications
has lead to a great deal of research and deve

metals with low diffusivity in solid aluminum are produced *in* $\frac{A}{A}$ elevated temperature, the dislocations are able to overcome *in* $\frac{B}{A}$ and $\frac{A}{A}$. The dispersoids by climb, but this mechanism alone is ins situ, such as Al-Fe-Ce alloys by Alcoa (PA, USA) and Al-Fe-
V-Si alloys developed by Allied Signal (NJ, USA). While these cient to explain the high strengths observed experimentally.^[3] materials have good elevated temperature strengths (Al-8Fe-
1.4V-1.7Si is reported as having a vield strength of 184 MPa laboratory tests into in-service conditions, it is necessary to 1.4V-1.7Si is reported as having a yield strength of 184 MPa laboratory tests into in-service conditions, it is necessary to at 315 °C^[1]), their use is limited by the hot workability. The have some type of physically b strength of the materials deteriorates in each thermomechanical behavior of the material. The model must allow the prediction ening at the high temperatures necessary for deformation. ters such as volume fraction of dispersoid, dispersoid size and

Following transmission electron microscopy (TEM) obserstable and insoluble in aluminum can be added to the materials. vations where dispersoids appeared to pin dislocations on the It also enables great flexibility in the design of materials with denarture side of the disperso It also enables great flexibility in the design of materials with departure side of the dispersoid as dislocations pass over them,^[4] chosen volume fractions, types, and sizes of dispersoids and Rösler and Arzt have dev chosen volume fractions, types, and sizes of dispersoids and Rösler and Arzt have developed the detachment model.^[5] The of alloying additions. rate controlling mechanism for this model is the detachment of a dislocation from the departure side of an attractive dispersoid. The attractive force results from the reduction in the elastic **Jane Minay, Richard Dashwood,** and **Henry McShane,** Department e.minay@ic.ac.uk. This occurs by diffusional relaxation at elevated temperatures,

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Table 1 Nominal compositions of materials tested

Alloy Composition (wt. %)	Ceramic Additions (Vol. %)	Ceramic Radius (nm)
$Al-1Mg-0.35Li$	None	n/a
Al	10% Al ₂ O ₃	13
Al	10% TiO ₂	23
$Al-1Mg-0.35Li$	10% Al ₂ O ₃	13
Al-1Mg-0.35Li	10% TiO ₂	23
Al-1Mg-0.35Li	7.5% TiO ₂	23

or by a reduction in the dislocation surface energy at the interface, and is represented by a factor *K*. Although the model fits some of the experimental data, it fails to represent the temperature dependence of deformation behavior and is extremely sensitive to the factor *K*, which cannot be measured experimentally.

The results of constant strain rate compression tests for a number of ultrafine dispersion-strengthened aluminum alloys are presented in this article. The effects of temperature, volume **Fig. 1** Stress strain behavior of Al + Al₂O₃ tested at 0.1 s⁻¹ fraction, strain rate, dispersoid composition, and alloying additions are demonstrated. The behavior is compared to the detachment model, which is found to be inadequate at representing
the material behavior. The detachment model has been adapted
by the introduction of a variable K value that provides an
improved correlation between the predicte improved correlation between the predicted and experimental

with different ceramic powders $(A_1O_3$ with a mean diameter (A_2O_3) and (A_1O_3) in the sample of 13 nm and TiO₂ with a mean diameter of 23 nm), followed An infra red (IR) furnace was used to heat the samples. Polyt

out phase identification of the ceramic powders and the extruded materials. The solid samples were spun during x-ray diffraction to avoid intensity reflections due to preferred orientations **3. Results and Discussion** resulting from the extrusion process. The samples were scanned from $2\theta = 20$ to 100° in steps of 0.04°. The *d*-spacings corres- **3.1 Phase Distribution** ponding to the resulting peaks were calculated using the Figure 2 shows part of the x-ray diffraction trace for the Figure 2 shows part of the x-ray diffraction trace for the TiO₂ powder added to the TiO₂ powder added

was performed in a JEOL 120CX microscope (Japan Electron properties.
Optics Ltd., Tokyo) at 100 kV.

2. Experimental 2.4 Mechanical Testing

Cylindrical compression samples measuring 10 mm in height **2.1 As-Received Materials** and 8 mm in diameter were machined from the rods parallel The materials studied were provided as consolidated billets
by AMC (Farnborough, United Kingdom). The fabrication pro-
cess involved mechanically alloying Al, Li, and Mg powders
with different ceramic powders (ALO, with a of 13 nm and TiO₂ with a mean diameter of 23 nm), followed
by hot isostatic pressing into billets. These billets were hot
extruded, at Imperial College, in an ENEFCO 5MN hydraulic
vertical press, into rods with a diamete **2.2** *X-Ray Diffraction* **2.2** *X-Ray Diffraction* **1 1 2.2** *X-Ray Diffraction* **1 2.2** *X-Ray Diffraction* **1** *Provide true stress and strain of 0.3. Typical 1 <i>Provide true stress and strain of 0.3. Ty* A computer-controlled Philips x-ray diffractometer (Philips results from the tests of the Al + Al₂O₃ material tested at a Electronic Instruments Corp., Mahwah, NJ), with a PW 1050/ strain rate of 0.1 s⁻¹ are shown i Electronic Instruments Corp., Mahwah, NJ), with a PW 1050/ strain rate of 0.1 s⁻¹ are shown in Fig. 1. After testing, the 25-model goniometer, was used with Cu K_a radiation to carry samples were water quenched to reta samples were water quenched to retain the dislocation structure.

2.3 Transmission Electron Microscopy 1.1 1.1 *Please and Please interprocess Please in the extruded mate- Please in the extruded mate- Please in the extruded mate-*Longitudinal and transverse sections were cut from the $\frac{1}{3}$ rial correspond to the Al₃Ti phase. The same behavior is extruded and deformed specimens and ground to a thickness observed in the materials mechanically observed in the materials mechanically alloyed with TiO₂ and

Fig. 2 X-ray diffraction spectra of TiO₂ powder used in the mechanical alloying process and the extruded alloys containing Ti

MgLi additions. The reaction $13Al + 3TiO₂ \rightarrow 2Al₂O₃ +$ 3TiAl3 leads to a large reduction in free energy of the system. The high temperatures used in the consolidation of the powders and the extrusion process result in diffusion rates sufficiently high so as not to impede the kinetics of the reaction.

Rösler *et al.*^[6] have observed the substitution of Al_2O_3 by MgO in mechanically alloyed materials of Al and Al_2O_3 containing Mg. This is not discernible in the materials currently being investigated due the proximity of any such peaks to the large Al peaks. Due to the small size of the dispersoids, it has not been possible to directly analyze the dispersoid compositions *via* TEM.

3.2 Microstructure

The microstructures of the materials have been studied using TEM. The materials alloyed with $TiO₂$ contain dispersoids approximately 25 nm in diameter, and the materials alloyed with Al_2O_3 powder contain dispersoids approximately 15 nm in diameter. Figure 3 is a micrograph of the AlLiMg + Al_2O_3 material in the as-extruded condition. It shows a relatively homogenous distribution of spherical dispersoids with few dislocations. A fine substructure has been developed during extrusion consolidation with a subgrain size of \sim 300 nm. Figure 4 shows a micrograph of the 7.5% TiO₂ material following deformation at 350 °C and a strain rate of 0.001 s^{-1} . There is a high density of dislocations. This may account for the work softening behavior illustrated in Fig. 1. It is proposed that the dispersoids can act as prolific sources of mobile dislocations from the incoherent dispersoid matrix interface, which result in softening during deformation.

Figure 5 is a micrograph of the same sample as Fig. 4. It contains a dislocation configuration that appears to show pinning of the dislocation on the departure side of a dispersoid. This is in support of the rate controlling mechanism of the **Fig. 5** TEM image of departure side pinning of a dislocation by detachment model. a dispersoid

Fig. 3 TEM image of AlLiMg $+$ Al₂O₃ as-extruded showing good distribution of dispersoid particles and fine substructure

Fig. 4 TEM image of AlLiMg $+ 7.5\%$ TiO₂ after deformation at 350 $^{\circ}$ C and 0.001 s⁻¹ showing high density of dislocations

Fig. 6 Mechanical behavior of mechanically alloyed Al-Mg-Li powder containing no ceramic dispersoid

Fig. 7 Mechanical behavior of Al-Mg-Li + 10% Al₂O₃

The results of the constant strain rate compression tests are shown in Fig. 6 to 11. All of the dispersion-strengthened mechanically alloyed, unreinforced material. atoms have no observable effect at elevated temperature.

The effect of the Mg-Li solute additions is considered in Fig. 12 and 13. From Fig. 12, it can be seen that the addition of Mg-Li to the material alloyed with $TiO₂$ has no effect. A

Fig. 8 Mechanical behavior of Al + 10% Al_2O_3

Fig. 9 Mechanical behavior of Al $+ 10\%$ TiO₂

small decrease is strength has been observed on alloying with **Mg-Li** additions and Al₂O₃ dispersoids at the highest tempera-
3.3 Mechanical Behavior tures and lower strains rates. A weakening effect has previously been observed by alloying $Al + Al₂O₃$ with $Mg³$. It would appear that this results from reaction between the $Al₂O₃$ dispermaterials show excellent elevated temperature properties with soids and Mg to change either the dispersoid size or the behavior
flow stresses greater than 150 MPa at 350 °C. The strength at of the dispersoid matrix interf flow stresses greater than 150 MPa at 350 °C. The strength at of the dispersoid matrix interface. No such reaction occurs with this temperature is more than doubled when compared to the the Al₃Ti dispersoids; thus, it m

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n = \frac{\partial \ln \dot{\varepsilon}}{\partial \ln \sigma}|_T \qquad (\text{Eq 1})
$$

Fig. 10 Mechanical behavior of Al-Mg-Li $+ 10\%$ TiO₂

Fig. 11 Mechanical behavior of Al-Mg-Li $+ 7.5\%$ TiO₂

high and to vary considerably with temperature, making tradi-
dence of the flow stress cannot be reproduced.

Fig. 12 Overlay of Fig. 9 (dashed lines) and 10 (solid lines)

Fig. 13 Overlay of Fig. 7 (solid lines) and 8 (dashed lines)

As observed in similar materials, $[7]$ *n* is found to be extremely rate to the stress is adequately predicted, the temperature depen-

tional deformation equations inappropriate for modeling the On further consideration, it would seem reasonable that *K* behavior. The *n* values for the materials $AI + 10\%$ AI_2O_3 , AI - might vary with temperature due to variations of dislocation Mg-Li + 10% AI_2O_3 , and AI -Mg-Li + 10% TiO₂ are plotted energy with temperature. The va Mg-Li + 10% Al₂O₃, and Al-Mg-Li + 10% TiO₂ are plotted energy with temperature. The value of *K* relates the energy of a dislocation in 203 and 203, and Al-Mg-Li + 10% TiO₂ are plotted a dislocation at a dispersoi a dislocation at a dispersoid to the energy of a dislocation in The behavior of the Al-Mg-Li + Al₂O₃ material has been the matrix. $E_{\text{Dispersoid}} = KE_{\text{Matrix}}$. As the temperature of the compared to the detachment model using a K value of 0.8, as material increases, more defects exist in t material increases, more defects exist in the matrix so the strain used previously for Al_2O_3 dispersoids by Rösler *et al.* ^[6] A energy of the dislocation in the matrix is reduced, increasing dislocation density of 10^{13} has been used and other variables the K value. The strain the K value. The strain energy at an incoherent interface would in the model have been taken to be the same as for Al. The be affected less by an increase in temperature. At higher temperresults are plotted in Fig. 15. While the sensitivity of the strain atures, there will also be more mobile dislocations. Increasing

Fig. 14 Variation of stress exponent with temperature for two of the materials tested

Fig. 15 Detachment model predictions (lines) and the experimental **Fig. 17** Effect of volume fraction in TiO₂ containing material and results for the Al-Mg-Li + 10% Al₂O₃ material and the adapted detachment model results for the Al-Mg-Li + 10% Al_2O_3 material

deformation stress. In order to fit the experimental data by

The fit is less satisfactory for the lowest temperature of 250 respectively) is best. 8C. At this temperature, the stress developed at the strain rates While the fit of the model with the experimental data is still

Fig. 16 Modified detachment model (lines) with variable *K* values and experimental results for the Al-Mg-Li + 10% Al_2O_3 material

the mobile dislocation density with temperature in the model different dispersoids. The values for *K* have been determined would also lead to an increased temperature dependence of the for the Al-Mg-Li + 10% TiO₂ material by fitting the detachment deformation stress. In order to fit the experimental data by model to the data. These K val changing dislocation density alone, it must increase by a factor model for a volume fraction of 7.5% of 23 nm radius dispersoids of more than 10^{20} , which would be unrealistic. and compared to the test results for the more than 10^{20} , which would be unrealistic. and compared to the test results for the 7.5% TiO₂ containing It is proposed that the ratio of the energy of the dislocation material. Figure 17 shows the result. The soli material. Figure 17 shows the result. The solid lines are those at the dispersoid to the energy at the matrix decreases with used to determine K values for the 10% TiO₂ containing matetemperature. By varying *K* linearly with temperature, a much rial. The dashed lines are the predicted results using these *K* improved fit with the experimental data is obtained. The fit of values for the 7.5% TiO₂ containing material. The results for the experimental data to the modified detachment model with temperatures of 350 to 550 °C are temperatures of 350 to 550 $^{\circ}$ C are shown, where the correlation a variable *K* value, for the same material, is shown in Fig. 16. to the experimental data (solid circles and hollow circles,

used is greater than the detachment stress, so this will no longer far from perfect, it is much improved on the original version be the rate controlling mechanism. $\qquad \qquad$ of the detachment model. The strain rate dependence and the The variation of *K* with temperature is different for the temperature dependence can be represented well using the model, and the prediction of the effect of volume fraction is directly, the new version of the model would seem to be the reasonably good. best currently available as a predictive tool for deformation

Acknowledgments 5. Conclusions

dispersion-strengthened materials tested appears to be extremely promising for elevated temperature applications. Tarrant (AMC, Farnborough, United Kingdom) for the provi-

Addition of solute atoms to these types of alloys seems to sion of materials. have little effect at the temperatures investigated, except when there is a reaction between the solute atoms and the dis- **References**

persed phases.

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The detachment model fails to predict the correct tempera-
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The detachment model fails to predict the correct tempera-

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⁴ V.C. Nardone and J.K. Tien: *Scripta Metall.*, 1983, vol. 17, p. 467. ture dependence of these materials. However, by varying the *A. V.C. Nardone and J.K. Tien: Scripta Metall.*, 1983, vol. 17, p. 467.
 K value in the detachment model linearly with temperature, a

much improved fit with e certain drawbacks of the detachment model still exist, for exam- 7. S.C. Khatri, A. Lawley, M.J. Koczak, and K.G. Grassett: *Mater. Sci.* ple, the fact that the *K* value cannot be measured or predicted *Eng.*, 1993, vol. A167, pp. 11-21.

behavior in these types of fine dispersoid-reinforced materials.

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